

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

# 4-[4-(4-Amino-1,2,5-oxadiazol-3-yl)-1,2,5-oxadiazol-3-yl]-1,2,5-oxadiazol-3-amine

 Si-Yuan Jia,<sup>a</sup> Bo-Zhou Wang,<sup>a</sup> Xue-Zhong Fan,<sup>a</sup> Ping Li<sup>b</sup> and Seik Weng Ng<sup>c,d\*</sup>

<sup>a</sup>Xi'an Modern Chemistry Research Institute, Xi'an 710065, People's Republic of China, <sup>b</sup>Department of Chemistry, Jining Teachers College, Wulanchabu 012000, Inner Mongolia, People's Republic of China, <sup>c</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and <sup>d</sup>Chemistry Department, King Abdulaziz University, PO Box 80203 Jeddah, Saudi Arabia  
Correspondence e-mail: seikweng@um.edu.my

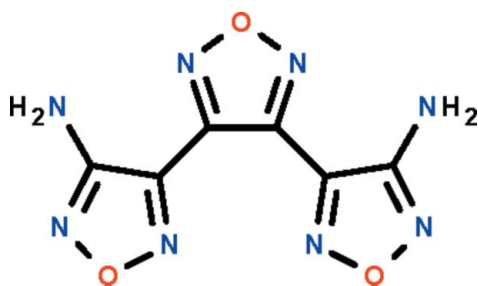
Received 19 April 2012; accepted 21 April 2012

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.031;  $wR$  factor = 0.096; data-to-parameter ratio = 12.0.

The complete molecule of the compound,  $\text{C}_6\text{H}_4\text{N}_8\text{O}_3$ , is generated by a crystallographic twofold rotation axis that runs through the central ring. The flanking ring is twisted by  $20.2(1)^\circ$  with respect to the central ring. One of the amino H atoms forms an intramolecular  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bond; adjacent molecules are linked by  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds forming a chain running along  $[10\bar{2}]$ .

## Related literature

For the synthesis, see: Kulikov & Kakhova (1994); Zhou *et al.* (2007).



## Experimental

## Crystal data

$\text{C}_6\text{H}_4\text{N}_8\text{O}_3$	$V = 931.9(2) \text{ \AA}^3$
$M_r = 236.17$	$Z = 4$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 7.1681(9) \text{ \AA}$	$\mu = 0.14 \text{ mm}^{-1}$
$b = 10.8147(13) \text{ \AA}$	$T = 293 \text{ K}$
$c = 12.3448(18) \text{ \AA}$	$0.33 \times 0.26 \times 0.17 \text{ mm}$
$\beta = 103.155(1)^\circ$	

## Data collection

Bruker SMART APEX diffractometer	1047 independent reflections
2675 measured reflections	933 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.014$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.031$	87 parameters
$wR(F^2) = 0.096$	All H-atom parameters refined
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.28 \text{ e \AA}^{-3}$
1047 reflections	$\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N4}-\text{H1}\cdots\text{N1}$	0.90 (2)	2.37 (2)	2.932 (2)	121 (1)
$\text{N4}-\text{H2}\cdots\text{N3}^i$	0.87 (2)	2.23 (2)	3.070 (2)	162 (2)

 Symmetry code: (i)  $-x + 1, -y + 1, -z + 1$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

We acknowledge support from the Equipment Department Preselected Project (grant No. 404060020502) and the Ministry of Higher Education of Malaysia (grant No. UM.C/HIR/MOHE/SC/12).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5881).

## References

- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.  
 Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Kulikov, A. S. & Kakhova, N. N. (1994). *Russ. Chem. Bull.* **43**, 630–632.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
 Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.  
 Zhou, Y.-S., Li, J.-K. & Huang, X.-P. (2007). *Chin. J. Explosives Propellants*, **30**, 454–556.

## supplementary materials

*Acta Cryst.* (2012). E68, o1573 [doi:10.1107/S1600536812017825]

## 4-[4-(4-Amino-1,2,5-oxadiazol-3-yl)-1,2,5-oxadiazol-3-yl]-1,2,5-oxadiazol-3-amine

Si-Yuan Jia, Bo-Zhou Wang, Xue-Zhong Fan, Ping Li and Seik Weng Ng

### Comment

We are interested in *N*-heterocyclic compounds having few hydrogen atoms as these compounds are a source of explosives. In the title compound (Scheme I), the hydrogen atoms constitute an amino group. In  $\text{NH}_2\text{-C}_2\text{N}_2\text{O-C}_2\text{N}_2\text{O-C}_2\text{N}_2\text{O-NH}_2$ , two amino-substituted 1,2,5-oxadiazole rings flanking a central 1,2,5-oxadiazole ring; the molecule lies on a twofold rotation axis that relates one flanking ring to the other (Fig. 1). The flanking ring is twisted by  $20.2(1)^\circ$  with respect to the central ring. One of the amino H atoms forms an intramolecular hydrogen bond; adjacent molecules are linked by an  $\text{N-H}\cdots\text{N}$  hydrogen bond (Table 1, Fig. 2). to form a chain running along  $[1\ 0\ -2]$ .

### Experimental

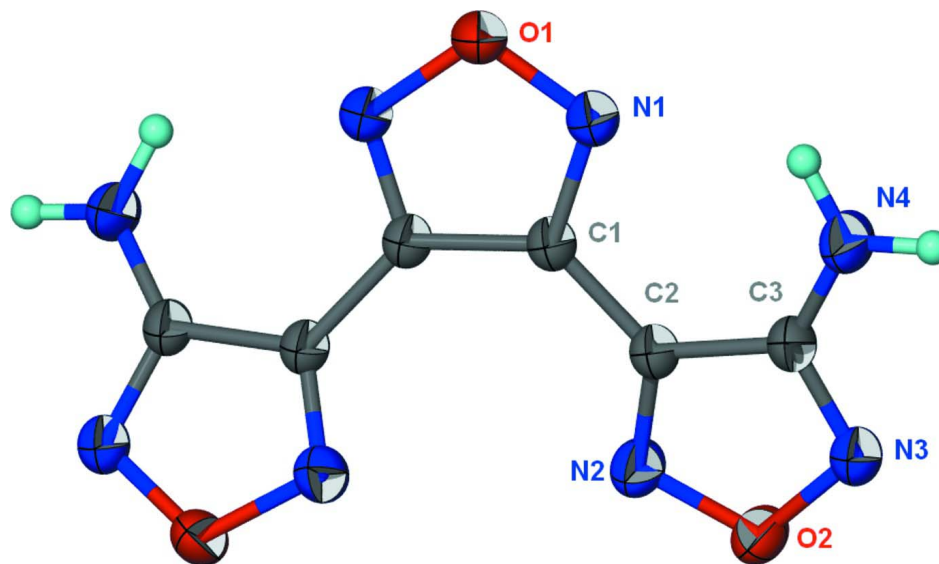
3,4-Bis(4'-aminofurazano-3')furoxan was synthesized by using a literature procedure (Zhou *et al.*, 2007). The compound (7.5 g) was dissolved in acetic acid (30 ml). The solution was added to a reducing agent prepared from stannous chloride dihydrate (22.6 g, 100 mmol) dissolved in acetic anhydride (20 ml), acetic acid (100 ml) and concentrated hydrochloric acid (20 ml). The reduction was performed according to an literature procedure (Kulikov & Kakhova, 1994). The mixture was heated at 348 K for 8 h. The cool mixture was then poured into water (150 ml). The white precipitate that separated was collected and recrystallized from an ethyl acetate/ether mixture; yield 70%, m.pt. 456–457 K. The purity was established by HPLC to be 99.6%. CH&N elemental analysis. Calculated for  $\text{C}_6\text{H}_4\text{N}_8\text{O}_3$  (%): C 30.51, N 47.46, H 1.69. Found: C 30.41, N 47.58, H 1.61.

### Refinement

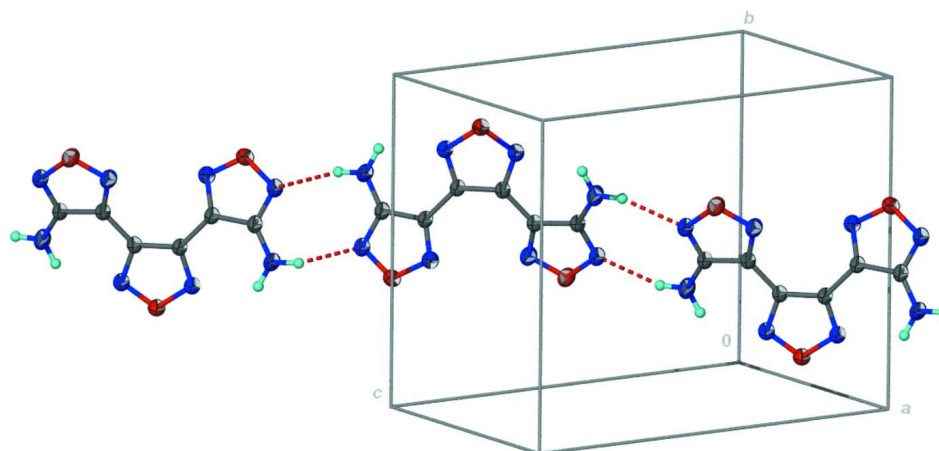
The H-atoms were located in a difference Fourier map, and were refined freely.

### Computing details

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINTE* (Bruker, 2009); data reduction: *SAINTE* (Bruker, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).


**Figure 1**

Anisotropic displacement ellipsoid plot (Barbour, 2001) of  $C_6H_4N_8O_3$  at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius. The molecule is located on a twofold rotation axis; symmetry-related atoms are not labeled.


**Figure 2**

Hydrogen-bonded chain structure. The intermolecular H bond is drawn as a dashed line, the intramolecular H bond is not shown.

#### 4-[4-(4-Amino-1,2,5-oxadiazol-3-yl)-1,2,5-oxadiazol-3-yl]-1,2,5-oxadiazol-3-amine

##### Crystal data

$C_6H_4N_8O_3$

$M_r = 236.17$

Monoclinic,  $C2/c$

Hall symbol:  $-C 2yc$

$a = 7.1681 (9) \text{ \AA}$

$b = 10.8147 (13) \text{ \AA}$

$c = 12.3448 (18) \text{ \AA}$

$\beta = 103.155 (1)^\circ$

$V = 931.9 (2) \text{ \AA}^3$

$Z = 4$

$F(000) = 480$

$D_x = 1.683 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 1575 reflections

$\theta = 3.4\text{--}27.7^\circ$

$\mu = 0.14 \text{ mm}^{-1}$

$T = 293$  K  $0.33 \times 0.26 \times 0.17$  mm  
 Prism, colorless

*Data collection*

Bruker SMART APEX 933 reflections with  $I > 2\sigma(I)$   
 diffractometer  $R_{\text{int}} = 0.014$   
 Radiation source: fine-focus sealed tube  $\theta_{\text{max}} = 27.5^\circ$ ,  $\theta_{\text{min}} = 3.4^\circ$   
 Graphite monochromator  $h = -9 \rightarrow 9$   
 $\omega$  scans  $k = -14 \rightarrow 13$   
 2675 measured reflections  $l = -15 \rightarrow 8$   
 1047 independent reflections

*Refinement*

Refinement on  $F^2$  Hydrogen site location: inferred from  
 Least-squares matrix: full neighbouring sites  
 $R[F^2 > 2\sigma(F^2)] = 0.031$  All H-atom parameters refined  
 $wR(F^2) = 0.096$   $w = 1/[\sigma^2(F_o^2) + (0.0566P)^2 + 0.219P]$   
 $S = 1.08$  where  $P = (F_o^2 + 2F_c^2)/3$   
 1047 reflections  $(\Delta/\sigma)_{\text{max}} < 0.001$   
 87 parameters  $\Delta\rho_{\text{max}} = 0.28 \text{ e } \text{\AA}^{-3}$   
 0 restraints  $\Delta\rho_{\text{min}} = -0.17 \text{ e } \text{\AA}^{-3}$   
 Primary atom site location: structure-invariant Extinction correction: *SHELXL97* (Sheldrick,  
 direct methods 2008),  $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$   
 Secondary atom site location: difference Fourier Extinction coefficient: 0.020 (3)  
 map

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.0000	0.80892 (10)	0.7500	0.0471 (3)
O2	0.16242 (12)	0.35531 (8)	0.56878 (8)	0.0482 (3)
N1	0.09563 (14)	0.73719 (9)	0.68904 (8)	0.0431 (3)
N2	0.06415 (14)	0.41475 (9)	0.63634 (9)	0.0446 (3)
N3	0.29963 (15)	0.43342 (9)	0.54042 (9)	0.0453 (3)
N4	0.3974 (2)	0.63687 (11)	0.58898 (12)	0.0626 (4)
H1	0.372 (2)	0.7101 (14)	0.6165 (12)	0.059 (4)*
H2	0.480 (3)	0.6342 (15)	0.5474 (15)	0.064 (5)*
C1	0.06039 (15)	0.62259 (9)	0.71097 (9)	0.0349 (3)
C2	0.13553 (15)	0.52528 (10)	0.65104 (9)	0.0358 (3)
C3	0.28563 (16)	0.53785 (10)	0.59138 (10)	0.0393 (3)

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0644 (8)	0.0324 (6)	0.0520 (7)	0.000	0.0291 (6)	0.000
O2	0.0540 (5)	0.0385 (5)	0.0600 (6)	-0.0051 (4)	0.0291 (4)	-0.0103 (4)
N1	0.0529 (6)	0.0356 (5)	0.0471 (6)	-0.0024 (4)	0.0248 (5)	-0.0011 (4)
N2	0.0460 (5)	0.0397 (5)	0.0549 (6)	-0.0046 (4)	0.0255 (5)	-0.0068 (4)
N3	0.0513 (6)	0.0401 (5)	0.0526 (6)	0.0004 (4)	0.0287 (5)	-0.0004 (4)
N4	0.0745 (8)	0.0423 (6)	0.0915 (10)	-0.0118 (5)	0.0614 (8)	-0.0088 (6)
C1	0.0364 (5)	0.0344 (5)	0.0373 (5)	-0.0010 (4)	0.0151 (4)	0.0007 (4)
C2	0.0374 (5)	0.0349 (6)	0.0390 (6)	-0.0004 (4)	0.0166 (4)	0.0010 (4)

C3      0.0435 (6)      0.0366 (6)      0.0436 (6)      0.0017 (4)      0.0223 (5)      0.0022 (4)

*Geometric parameters (Å, °)*

O1—N1 <sup>i</sup>	1.3685 (11)	N4—C3	1.3419 (16)
O1—N1	1.3686 (11)	N4—H1	0.896 (16)
O2—N2	1.3684 (12)	N4—H2	0.871 (19)
O2—N3	1.4001 (13)	C1—C1 <sup>i</sup>	1.434 (2)
N1—C1	1.3055 (14)	C1—C2	1.4582 (15)
N2—C2	1.2967 (15)	C2—C3	1.4419 (15)
N3—C3	1.3077 (15)		
N1 <sup>i</sup> —O1—N1	110.94 (11)	N1—C1—C2	117.95 (9)
N2—O2—N3	110.93 (8)	C1 <sup>i</sup> —C1—C2	133.62 (6)
C1—N1—O1	106.22 (9)	N2—C2—C3	109.39 (10)
C2—N2—O2	106.07 (9)	N2—C2—C1	123.83 (9)
C3—N3—O2	105.40 (9)	C3—C2—C1	126.57 (10)
C3—N4—H1	121.5 (10)	N3—C3—N4	124.54 (11)
C3—N4—H2	118.5 (11)	N3—C3—C2	108.19 (10)
H1—N4—H2	118.6 (15)	N4—C3—C2	127.25 (11)
N1—C1—C1 <sup>i</sup>	108.31 (6)		
N1 <sup>i</sup> —O1—N1—C1	0.18 (6)	N1—C1—C2—C3	17.47 (17)
N3—O2—N2—C2	-0.32 (13)	C1 <sup>i</sup> —C1—C2—C3	-167.05 (15)
N2—O2—N3—C3	0.77 (13)	O2—N3—C3—N4	177.65 (12)
O1—N1—C1—C1 <sup>i</sup>	-0.43 (14)	O2—N3—C3—C2	-0.87 (13)
O1—N1—C1—C2	176.12 (8)	N2—C2—C3—N3	0.73 (14)
O2—N2—C2—C3	-0.23 (13)	C1—C2—C3—N3	-174.16 (11)
O2—N2—C2—C1	174.83 (10)	N2—C2—C3—N4	-177.73 (13)
N1—C1—C2—N2	-156.72 (11)	C1—C2—C3—N4	7.4 (2)
C1 <sup>i</sup> —C1—C2—N2	18.8 (2)		

Symmetry code: (i)  $-x, y, -z+3/2$ .

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N4—H1...N1	0.90 (2)	2.37 (2)	2.932 (2)	121 (1)
N4—H2...N3 <sup>ii</sup>	0.87 (2)	2.23 (2)	3.070 (2)	162 (2)

Symmetry code: (ii)  $-x+1, -y+1, -z+1$ .